

SUBSTITUTE SPECIFICATION

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METHOD AND APPARATUS FOR TREATING EXHAUST GAS

BACKGROUND OF THE INVENTION

Technical Field

5 The present invention relates to a method and apparatus for treating an exhaust gas, and more particularly to a method and apparatus for efficiently detoxifying an exhaust gas containing a fluorine compound which is discharged from a semiconductor fabrication process such as a dry-cleaning
10 process of an inner surface of a semiconductor fabrication apparatus or an etching process of various types of films such as oxide films.

Background Art

15 In a semiconductor fabrication process such as an etching process or a chemical vapor deposition (CVD) process, fluorine compounds such as hydrofluorocarbons (e.g., CHF₃) or perfluorocompounds (e.g., CF₄, C₂F₆, C₃F₈, C₄F₈, C₅F₈, C₄F₆, SF₆, and NF₃) may have been used in an apparatus. In some cases,
20 CO, NH₃ or O₂ may be used in a semiconductor fabrication apparatus. An exhaust gas discharged from a semiconductor fabrication apparatus which uses fluorine compounds, CO or NH₃ includes harmful components such as CO, NH₃, SiF₄, F₂, COF₂, C₅F₈, C₄F₆, or NF₃. The exhaust gas also includes fluorine compounds
25 which are not harmful but affect the global warming. Therefore, when an exhaust gas is discharged from a semiconductor fabrication apparatus using fluorine compounds, or the like environmental atmosphere, it is necessary to detoxify harmful gas included in the exhaust gas and also to decompose gas which
30 causes global warming.

In a conventional method of treating a harmful gas (SiF₄,

F_2 , COF_2 , C_5F_8 , C_4F_6 , or NH_3) included in an exhaust gas, harmful components are adsorbed by an adsorbent such as a synthetic zeolite. However, in this conventional method, perfluorocompounds (PFCS) cannot be removed from the exhaust 5 gas. Further, the adsorbent needs to be replaced with a new one periodically, resulting in increased running cost.

A wet treatment apparatus (scrubbing process) has been used to scrub an exhaust gas for removing a water-soluble gas and a hydrolytic gas such as SiF_4 , F_2 or NH_3 from the exhaust 10 gas. However, such a scrubbing process cannot remove gases that are not water-soluble, such as PFCS.

A method has been proposed of removing PFCS from an exhaust gas with various types of catalysts for decomposing PFCS. However, if the catalyst is deteriorated, then harmful 15 components such as CO , C_5F_8 , and C_4F_6 may be discharged into the atmosphere immediately after deterioration of the catalyst. A method has also been proposed of treating PFCS by combustion. However, NO_x or CO may be produced as a by-product gas, depending on the combustion conditions. Since this method 20 requires a fuel such as H_2 , natural gas (city gas), or propane gas, it is necessary to provide equipment for supplying the fuel. Further, a complicated process is required to manage the operation. A method has also been proposed of decomposing PFCS by a heating oxidative decomposition. However, in order 25 to decompose PFCS (e.g., CF_4) that are unlikely to otherwise be decomposed, an exhaust gas should be heated to a temperature of $1400^\circ C$ or higher. In such a case, on the requirements of materials and a heater in the system become considerable.

A method has been proposed in which NH_3 , a lower saturated 30 hydrocarbon gas, or a lower unsaturated hydrocarbon gas is added to an exhaust gas, and PFCS are decomposed by a heating

oxidative decomposition without free O₂ gas. Further, there has also been proposed a method of decomposing PFC with a plasma in the presence of water (H₂O). However, when PFC is decomposed, a harmful gas such as CO or HF is produced and thermal NOx is
5 also produced. Therefore, it is necessary to provide a separate exhaust gas treatment apparatus for treating the harmful gas and the thermal NOx.

The Japanese patent No. 3217034 discloses a method in which water is added to an exhaust gas containing PFC and then
10 the exhaust gas is heated in the presence of oxygen to react with a catalyst so that the PFCS are decomposed or oxidized. However, there exist various types of fluorine compounds such as hydrofluorocarbons (e.g., CHF₃) and perfluorocompunds (e.g., CF₄, C₂F₆, C₃F₈, C₄F₈, C₅F₈, C₄F₆, SF₆, and NF₃). When such fluorine
15 compounds are heated in the presence of water, most of these fluorine compounds are decomposed and hence an acid gas such as hydrofluoric acid is produced when being heated. Consequently, there arises a problem in that a heating section of a gas treatment tank is corroded by an acid gas such as
20 hydrofluoric acid.

BRIEF SUMMARY OF THE INVENTION

The present invention has been made in view of the above drawbacks. It is therefore an object of the present invention
25 to provide a method and apparatus for treating an exhaust gas which does not produce an acid gas such as hydrofluoric acid at a heating section of a gas treatment tank, and can thus prevent the heating section of the gas treatment tank from being corroded by the acid gas.

30 According to one aspect of the present invention, there is provided a method of treating an exhaust gas containing a

fluorine compound, the method comprising: heating the exhaust gas in the presence of O₂; and then adding H₂O to the exhaust gas to decompose or oxidize the fluorine compound.

In a preferred aspect of the present invention, the 5 fluorine compound is decomposed or oxidized in the presence of a catalyst after H₂O is added to the exhaust gas.

In a preferred aspect of the present invention, the method of treating an exhaust gas further comprises: before the heating, removing at least one of a powdery component, a 10 water-soluble component, and a hydrolytic component from the exhaust gas.

In a preferred aspect of the present invention, the method of treating an exhaust gas further comprises: after the fluorine compound is decomposed or oxidized, removing an acid 15 gas, which is produced when the fluorine compound is decomposed, from the exhaust gas.

According to the present invention described above, an exhaust gas containing a fluorine compound is heated to a required temperature ranging from 600 to 900°C, for example, 20 by a heating section in an O₂ atmosphere without adding water to the exhaust gas. Therefore, since water does not exist in the heating section, an acid gas such as hydrofluoric acid is not produced even when the fluorine compound is decomposed or oxidized. Accordingly, the heating section is prevented from 25 being corroded by a hydrofluoric acid gas. O₂ is heated sufficiently by the heating section after being added to the exhaust gas, and then water is added to the exhaust gas in an H₂O adding section. Therefore, the fluorine compound is decomposed or oxidized efficiently on the downstream side of 30 the heating section. An acid gas produced when the fluorine compound is decomposed is easily removed from the exhaust gas

by an acid gas removal section disposed downstream of the H₂O adding section where the fluorine compound is decomposed or oxidized. In this manner, the exhaust gas is treated to be harmless, and is then released into the atmosphere.

5 In the present invention, air or an O₂ gas is supplied to the exhaust gas containing the fluorine compounds so as to add O₂ to the exhaust gas, and the exhaust gas is then heated to a temperature ranging from 600 to 900 °C while adding water (H₂O) to the exhaust gas. Thus, most of the fluorine compounds
10 except for PFCS such as CF₄ or SF₆, which are unlikely to be decomposed, can be decomposed or oxidized to be harmless. Further, decomposition or oxidization is performed in the presence of a catalyst, so that PFCS (e.g., CF₄, SF₆) that are unlikely to be decomposed can completely be decomposed at a
15 relatively low temperature of 600 to 900°C by a catalytic reaction. Therefore, when a catalytic reaction is performed after the heating oxidative decomposition, for example, gas components such as CF₄ and SF₆ that are unlikely to be decomposed but need to be treated in view of the global warming can
20 completely be decomposed and removed.

According to another aspect of the present invention, a method is provided of treating an exhaust gas containing a fluorine compound, the method comprising: heating the exhaust gas in the presence of O₂; and then adding H₂ to the exhaust gas to decompose or oxidize the fluorine compound.
25

In a preferred aspect of the present invention, the fluorine compound is decomposed or oxidized in the presence of a catalyst after H₂ is added to the exhaust gas.

30 In a preferred aspect of the present invention, the method of treating an exhaust gas further comprises: before the heating, removing at least one of a powdery component, a

water-soluble component, and a hydrolytic component from the exhaust gas.

In a preferred aspect of the present invention, the method of treating an exhaust gas further comprises: after the 5 fluorine compound is decomposed or oxidized, removing an acid gas, which is produced when the fluorine compound is decomposed, from the exhaust gas.

According to the present invention, a H₂ gas and an O₂ gas (e.g., air) are added to the exhaust gas, instead of adding 10 water (H₂O). When H₂ and O₂ are added to the exhaust gas, water can be produced in the exhaust gas which is heated to a high temperature. Such water (H₂O) produced from H₂ and O₂ can decompose or oxidize the fluorine compound as with normal water. If the exhaust gas contains oxygen (O₂) therein, water can be 15 produced by supplying only a H₂ gas to the exhaust gas.

According to another aspect of the present invention, an apparatus is provided for treating an exhaust gas containing a fluorine compound, the apparatus comprising: a heating section for heating the exhaust gas; an exhaust gas supply for 20 supplying the exhaust gas to the heating section; an H₂O adding section located just downstream of the heating section for adding H₂O to the exhaust gas by supplying H₂O or H₂ to the exhaust gas; and an acid gas removal section for removing an acid gas produced by a reaction between the exhaust gas and 25 H₂O.

In a preferred aspect of the present invention, the heating section comprises a heating wire, the heating wire is wound thickly at an inlet-side of the heating section, and is wound thinly at an outlet-side of the heating section.

30 In a preferred aspect of the present invention, the apparatus for treating an exhaust gas further comprises: a

catalytic reactor disposed downstream of the H₂O adding section for decomposing the fluorine compound by catalytic reaction.

It is preferable that the heating wire is wound thickly on two-thirds of the heating section, and is wound thinly on 5 the remaining one-third of the heating section. With this structure, it is possible to widen the temperature range necessary for the heating oxidative decomposition within the range of 600 to 900 °C. Further, it is also possible to prevent a temperature of the exhaust gas flowing into the subsequent 10 catalytic reactor from being increased to 900 °C or higher.

If the heating wire is wound thickly on the whole heating section, the temperature range necessary for the heating oxidative decomposition can be secured as with the case of the above structure. However, in this case, a temperature of the 15 exhaust gas flowing into the catalytic reactor may be increased to 900 °C or higher, thus causing the deterioration of the catalyst to be accelerated.

In a preferred aspect of the present invention, the apparatus for treating an exhaust gas further comprises: a 20 water heating pipe disposed at the heating section; wherein H₂O to be added to the exhaust gas in the H₂O adding section is supplied through the water heating pipe and is heated by the water heating pipe.

In a preferred aspect of the present invention, the 25 apparatus for treating an exhaust gas further comprises: a water heating pipe disposed outside of the heating section; and an external heater disposed on the water heating pipe; wherein H₂O to be added to the exhaust gas in the H₂O adding section is supplied through the water heating pipe and is 30 heated by the external heater.

According to the present invention, water (H₂O) to be

added to the exhaust gas in the H₂O adding section is heated sufficiently by the water heating pipe provided at the heating section, and is thus vaporized. Therefore, water to be added in the H₂O adding section is heated to a high temperature, and
5 hence the fluorine compound contained in the exhaust gas can efficiently be decomposed or oxidized. The water heating pipe may be disposed inwardly of a circumferential wall of the heating section in such a state that the water heating pipe extends linearly or spirally. Alternatively, the heating
10 section may have a double-wall structure comprising an inner wall and an outer wall, so that a heater is provided in the inner wall and the water heating pipe is provided in the outer wall. Water (H₂O) to be added in the H₂O adding section may be supplied from the water heating pipe extending from outside
15 to inside of the heating section, and may be heated by the external heater. With this structure, a temperature of the water can be controlled by the external heater as desired.

In a preferred aspect of the present invention, the apparatus for treating an exhaust gas further comprises: an
20 air ejector for maintaining a pressure of the exhaust gas, which has been treated by the heating section, the H₂O adding section, and the acid gas removal section, at a predetermined value; and a bypass pipe for returning a part of the treated exhaust gas to an inlet side of the apparatus so as to mix the
25 treated exhaust gas with the untreated exhaust gas.

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 is a block diagram showing an exhaust gas treatment
30 apparatus according to a first embodiment of the present invention;

FIG. 2A is a view showing an example of a gas treatment tank;

FIG. 2B is a schematic enlarged view showing a heating wire incorporated in the gas treatment tank shown in FIG. 2A;

5 FIG. 3 is a block diagram showing an exhaust gas treatment apparatus according to a second embodiment of the present invention;

FIG. 4A through FIG. 4D are views each showing an example of a water (H_2O) supply or a H_2 supply incorporated in the gas 10 treatment tank; and

FIG. 5 is a block diagram showing an exhaust gas treatment apparatus according to a third embodiment of the present invention.

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DETAILED DESCRIPTION OF THE INVENTION

An exhaust gas treatment apparatus according to embodiments of the present invention will be described below with reference to the drawings. Like or corresponding parts 20 are denoted by like or corresponding reference numerals throughout drawings and will not be described below repetitively.

FIG. 1 shows an exhaust gas treatment apparatus according to a first embodiment of the present invention. The exhaust 25 gas treatment apparatus comprises a pre-treatment section for removing powdery components, water-soluble components, or hydrolytic components from an exhaust gas containing fluorine compounds, a heating oxidative decomposing section for performing heating oxidative decomposition of the pre-treated 30 exhaust gas, and a post-treatment section for post-treating an acid gas such as HF which has been produced by the heating

oxidative decomposition.

In this embodiment, a fan scrubber 1 serves as the pre-treatment section. An exhaust gas to be treated passes through the fan scrubber 1 and then passes through a mist separator 2. The exhaust gas is introduced from the mist separator 2 into a gas treatment tank 3 through an exhaust gas introduction pipe 11. The fan scrubber 1 and the mist separator 2 are supplied with service water or industrial water, which is sprayed from a water spray provided in each of the fan scrubber 1 and the mist separator 2. The exhaust gas containing fluorine compounds flows from the fan scrubber 1 toward the mist separator 2 while being brought into contact with water sprayed from the water sprays, and hence powdery components, water-soluble components, or hydrolytic components are removed from the exhaust gas. Instead of the fan scrubber 1, the pre-treatment section may comprise a water spray tower, a gas passage stirring tank, or an adsorption tank filled with an adsorbent such as zeolite or activated carbon. The water spray tower, the gas passage stirring tank, and the adsorption tank can also achieve the aforementioned effects. Water is retained in a circulating tank 15a, and is supplied to the fan scrubber 1 by a circulating pump 15 which compresses the water.

Devices to be used as the pre-treatment section can properly be selected according to components in the exhaust gas to be treated, a degree to which powders are mixed, or states of the plant. Since an adsorption tank filled with an adsorbent uses no water, it is suitable for a case where no waste water treatment equipment is provided in the system. If the exhaust gas contains materials which should specially be maintained, such as arsenic (As) or lead (Pb), then waste water used in the wet scrubbing process is contaminated by such

materials. For such cases, the wet scrubbing process is not suitable, but the adsorption process is suitable. The fan scrubber has a high rate of removal of the components with a small amount of water being supplied. Further, the fan scrubber has an excellent capability of capturing powders. Although the water spray tower requires an increased amount of water to be supplied in order to achieve a high rate of removal of the components, the water spray tower can reduce cost of the apparatus because of its simple structure. A liquid in the gas passage stirring tank can be adjusted in pH by a neutralizing liquid supplied thereto. Therefore, the gas passage stirring tank has a high rate of removal of components that would be difficult to be removed by another wet scrubbing process.

In the pre-treatment section, as described above, powdery components, water-soluble components, or hydrolytic components are removed from the exhaust gas with water or an adsorbent. For example, an acid gas such as SiF₄ or F₂ is removed from the exhaust gas in the pre-treatment section.

The exhaust gas which has passed through the pre-treatment section, i.e. the fan scrubber 1, is introduced into the gas treatment tank 3 through the exhaust gas introduction pipe 11, and is then heated to be oxidized or decomposed. The heating oxidative decomposition is performed in the gas treatment tank 3. An air pipe (i.e., an O₂ supply pipe) 5 is connected to the gas treatment tank 3 for supplying O₂ necessary for the reaction. A water pipe 6 is connected to the gas treatment tank 3 for supplying water (e.g., city water or industrial water) necessary for the reaction. The city water or the industrial water is purified up to a level of distilled water or pure water by a water purifier 7 connected

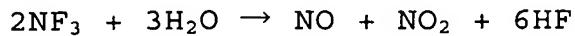
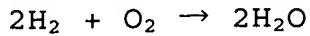
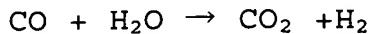
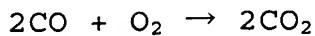
to the water pipe 6, and is then supplied to the gas treatment tank 3. Thereafter, the purified water is vaporized by a water heating pipe and is supplied to an end portion of a heating section of the gas treatment tank 3. Alternatively, water required for the reaction may be sprayed with a sprayer provided in the heating oxidative decomposing tank (the gas treatment tank 3) without being vaporized by the water heating pipe. In this case, the sprayed water is heated and vaporized in the heating oxidative decomposing tank.

FIG. 2A shows an example of a structure of the gas treatment tank, and FIG. 2B is a schematic enlarged view showing a heating wire incorporated in the gas treatment tank shown in FIG. 2A. The gas treatment tank 3 has an electric tube furnace 9 made of ceramics and serving as a heater. A large number of plate members 10 having an excellent thermal conductivity are provided on an inner surface of the electric tube furnace 9 to form a detour path through which the exhaust gas passes. As shown in FIG. 2B, the electric tube furnace 9 has a heating wire 9a which is wound thickly on two-thirds of the heating section 30 and is wound thinly on the remaining one-third of the heating section 30. With this structure, a mixture of the exhaust gas containing fluorine compounds and air (O_2) supplied from the air pipe 5 is heated to a temperature ranging from 700 to 900°C. The manner in which the heating wire 9a is arranged is not limited to the manner shown in FIG. 2B. For example, the heating wire 9a may be wound twice or more at the inlet-side of the heating section 30 and maybe wound once at the outlet-side of the heating section 30. A water heating pipe 8 is provided in the gas treatment tank 3, and is connected to the water pipe 6 which supplies water (H_2O). The water heating pipe 8 extends linearly from an inlet portion

to an end portion of a heating section (a region 30 shown in FIG. 2A) comprising the electric tube furnace 9 and the plate members 10. Water (H_2O) supplied from the water pipe 6 is heated and vaporized by the water heating pipe 8 to be formed into 5 a high-temperature vapor. The water as a high-temperature vapor is ejected from an open end of the water heating pipe 8 to the exhaust gas which has been heated to a high temperature, and is mixed with and added to the exhaust gas. If necessary, an oxygen supply communicating with the exhaust gas 10 introduction pipe 11 (the exhaust gas supply) or the heating section 30 may be provided for supplying oxygen (e.g., air, oxygen-rich air, pure oxygen, or ozone).

Thus, in an H_2O adding section (i.e., a region 40 shown in FIG. 2A) located just downstream of the heating section 30 15 of the gas treatment tank 3, oxidation of CO and decomposition of PFCS having four or more carbon atoms, hydrofluorocarbons, and NF_3 , are performed according to the following reaction formulas. The H_2O adding section 40 which is located downstream 20 of the heating section 30 is surrounded by a heat insulator provided in the gas treatment tank 3. In this H_2O adding section 40, the following reactions proceed due to addition 25 of water (H_2O) and O_2 to the exhaust gas which has been heated to a high temperature. Therefore, all of the components in the exhaust gas that are considered to be harmful to human bodies can be oxidized or decomposed. Although oxidation and 30 decomposition of fluorine compounds proceed even in the heating section 30, no hydrofluoric acid (HF) gas is produced in the heating section 30 because water (H_2O) is not added to the exhaust gas in the heating section 30. Therefore, in the gas treatment tank 3, the electric tube furnace 9 and the plate members 10 constituting the heating section 30 are not corroded

and damaged by a hydrofluoric acid (HF) gas.



10 In the oxidation, O_2 may be supplied from any O_2 sources such as atmospheric air, O_2 -rich air, and pure O_2 . Peroxide may be used as O_2 . In this gas treatment tank 3, the air pipe 5 is connected to the exhaust gas introduction pipe 11, so that air and the exhaust gas to be treated are introduced into the 15 gas treatment tank 3 from the top of the gas treatment tank 3. The aforementioned components in the exhaust gas can be oxidized and decomposed at a temperature ranging from 700 to 900°C. If the oxidation and decomposition are performed at a temperature of 1000°C or higher, then an amount of thermal 20 NO_x caused by N_2 in the air is increased. When the oxidation and decomposition are performed at a temperature of 900°C or lower, it is possible to economically select a fire-resistant element which is to be used in the gas treatment tank 3, and hence a manufacturing cost can be lowered. Therefore, in the 25 present embodiment, the oxidation and decomposition are performed at a temperature ranging from 700 to 900°C.

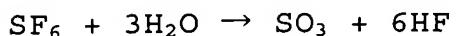
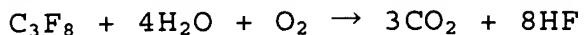
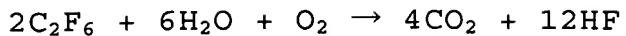
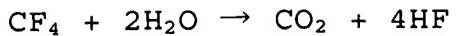
The post-treatment section post-treats an acid gas such as HF which is produced in the decomposing process of fluorine compounds. In the present embodiment, the post-treatment 30 section comprises a spray nozzle 13 serving as an acid gas removal section. The decomposed exhaust gas passes through

a water film formed by the spray nozzle 13 which is provided in the gas treatment tank 3, and the exhaust gas is then introduced to a mist separator 14 through an exhaust gas introduction pipe 11a. The exhaust gas passes through the mist 5 separator 14, and is then released as a harmless gas to the atmosphere. The spray nozzle 13 and the mist separator 14 are supplied with city water or industrial water, which is sprayed in each of the spray nozzle 13 and the mist separator 14. The exhaust gas is brought into contact with the water sprayed from 10 the spray nozzle 13. As a result, hydrofluoric acid (HF) gas produced by the decomposition of PFCS in the H₂O adding section (i.e., the decomposing treatment section) 40 of the gas treatment tank 3 is removed from the exhaust gas. Instead of the spray nozzle 13, the post-treatment section may comprise 15 a fan scrubber, a water spray tower, a gas passage stirring tank, or an adsorption tank filled with an adsorbent such as zeolite or activated carbon. The fan scrubber, the water spray tower, the gas passage stirring tank, and the adsorption tank can also achieve the aforementioned effects.

20 FIG. 3 shows an exhaust gas treatment apparatus according to a second embodiment of the present invention. As shown in FIG. 3, the gas treatment tank 3 further comprises a catalytic reactor 4 disposed downstream of the heating section 30. In the gas treatment tank 3, the exhaust gas is heated to be 25 oxidized or decomposed in the heating section 30, and is subsequently introduced into the catalytic reactor 4 where the exhaust gas is decomposed by catalytic reaction. The components of this embodiment other than the catalytic reactor 4 provided downstream of the H₂O adding section (i.e., the oxidative composition treatment section) 40 are the same as 30 the components of the exhaust gas treatment apparatus shown

in FIG. 1.

The catalytic reactor 4 has a catalyst therein for decomposing PFCS. The exhaust gas is introduced into an upper portion of the catalytic reactor 4 and flows downward from an upper catalyst layer to a lower catalyst layer. The electric tube furnace (i.e., heater) 9 made of ceramics is disposed upstream of the catalytic reactor 4, and hence the catalytic reactor 4 is heated to a temperature ranging from 600 to 900°C. In the catalytic reactor 4, a temperature of the catalyst can be maintained at a suitable level for catalytic reaction by heating the exhaust gas introduced into the catalytic reactor 4. Therefore, it is not necessary to provide a special device for heating the catalyst in the catalytic reactor 4. PFCS having three or less carbon atoms and SF₆ are brought into contact with the catalyst, so that the decomposition of PFCS and SF₆ is performed according to the following reaction formulas. In the following reaction formulas, O₂ which contributes to the decomposition has been added to the exhaust gas, to be treated, in the gas treatment tank 3, and water (H₂O) has been introduced into the exhaust gas at the end portion of the heating section 30 disposed upstream of the catalytic reactor 4. The catalyst comprises a catalyst for fluorine compounds, such as γ alumina or alumina zirconium composite material carried with tungsten oxide. With this structure, it is possible to decompose PFCS and SF₆ that are unlikely to be decomposed and have not been decomposed by the heat oxidative decomposition. Therefore, when the catalytic reactor 4 is incorporated into the exhaust gas treatment apparatus, components that do not directly affect humans but adversely affect global warming can be completely removed from an exhaust gas.



5 Next, modifications of the exhaust gas treatment apparatus according to the above embodiments of the present invention will be described. FIG. 4A shows a gas treatment tank 3 having a water heating pipe 8a which is disposed in the electric tube furnace 9 and extends spirally, instead of the
10 water heating pipe 8 (see FIG. 2A) which is disposed at the upper portion of the gas treatment tank 3 and extends linearly. With this structure, heat exchange between the exhaust gas and water (H_2O) flowing through the water heating pipe 8a is sufficient to produce a high-temperature vapor (H_2O), and hence
15 the high-temperature vapor (H_2O) can be added to the exhaust gas at the end portion of the heating section 30.

FIG. 4B shows a gas treatment tank 3 having a water heating pipe 8b which is disposed on the outer circumferential portion of the electric tube furnace 9 made of ceramics,
20 instead of the water heating pipes 8, 8a which are disposed in a gas passage formed in the electric tube furnace 9 made of ceramics. The water heating pipe 8b opens at the end portion of the heating section 30. The water heating pipe 8b is connected to a water pipe 6 for supplying city water or
25 industrial water, and is disposed outside of the electric tube furnace 9. Water (H_2O) flowing through the water heating pipe 8b is heated by exhaust heat of the electric tube furnace 9, and is then supplied to the end portion of the heating section 30 in the gas treatment tank 3. With this structure, a
30 high-temperature vapor (H_2O) heated by utilizing the exhaust heat of the electric tube furnace 9 can be supplied to the end

portion of the heating section 30.

FIG. 4C shows a gas treatment tank having a water heating pipe 8c which extends from outside to inside of the gas treatment tank 3 and opens at the end portion of the heating section 30 in the gas treatment tank 3, instead of the water heating pipes 8, 8a and 8b which are disposed inside or outside of the electric tube furnace 9. The water heating pipe 8c is connected to a water pipe 6 for supplying city water or industrial water through an external heater 16. Water (H_2O) supplied from the water pipe 6 is heated by the external heater 16 and then supplied to the end portion of the heating section 30 in the gas treatment tank 3. With this structure, water (H_2O) is heated to a suitable temperature without being affected by the electric tube furnace 9, and is then supplied to the end portion of the heating section 30. Water (H_2O) may be sprayed directly to the end portion of the heating section 30 without passing through the external heater 16, depending on treatment conditions.

FIG. 4D shows a gas treatment tank having a H_2 gas supply pipe 8d instead of the pipes for supplying water (H_2O). The H_2 gas supply pipe 8d opens at the end portion of the heating section 30 in the gas treatment tank 3. A H_2 gas supplied from the H_2 gas supply pipe 8d and an O_2 gas which has been added in advance to the exhaust gas are bonded together to produce water (H_2O). The water (H_2O) produced from the H_2 gas and the O_2 gas contains less impurities, compared to city water or industrial water. Therefore, it is possible to dispense with any equipment for distilling or purifying city water or industrial water, thereby reducing a cost of the treatment, apparatus as a whole.

Water (H_2O) is required for the oxidation of CO and the

decomposition of PFCS. Water (H_2O) is introduced into the exhaust gas treatment system in a vaporized state. If the water contains Si or Ca, then Si or Ca may be deposited or scales may be produced when the water (H_2O) is vaporized, thus causing
5 the apparatus to be clogged. Further, Cl contained in the water may cause a catalyst to deteriorate. Therefore, the supplied water (H_2O) needs to be purified water such as pure water or distilled water. Accordingly, it is necessary to provide an independent pipe for supplying pure water or distilled water
10 to the apparatus, and also to provide a device for producing pure water or a device for producing distilled water separately. Consequently, the installation cost of the apparatus is increased.
15 In the present embodiment, since water to be supplied is produced from a H_2 gas and an O_2 gas, the installation cost for providing pipes and the like can be eliminated.

FIG. 5 shows an exhaust gas treatment apparatus according to a third embodiment of the present invention. In this embodiment, an air ejector 16 is provided for maintaining a
20 pressure of an exhaust gas treated by the gas treatment tank 3 at a predetermined value. The air ejector 16 serves to discharge the exhaust gas forcibly from the gas treatment tank 3 where the exhaust gas passes therethrough, so that an internal pressure of the gas treatment tank 3 can be adjusted.
25 With this structure, the heating oxidative decomposition of the exhaust gas in the gas treatment tank 3 can be performed at a suitable pressure. Further, an analyzer 17 is provided for analyzing a concentration of components contained in the treated exhaust gas and a temperature of the treated exhaust
30 gas. Furthermore, a bypass pipe 18 and a bypass valve 19 are also provided for returning a part of the treated exhaust gas

to the inlet side of the fan scrubber 1 so as to mix the treated exhaust gas with the untreated exhaust gas. With this structure, the exhaust gas can be treated repetitively, and the exhaust gas can also be easily treated in a case where the 5 heating oxidative decomposition is not required. Other components of the exhaust gas treatment apparatus according to this embodiment are the same as those of the exhaust gas treatment apparatus according to above-mentioned embodiments.

Next, test results of exhaust gas treatment with a 10 testing apparatus equivalent to the exhaust gas treatment apparatus according to the second embodiment will be described. An N₂ gas mixed with various types of gases to be treated was introduced into the testing apparatus, and the concentration (ppm) of components in the gas was measured at a plurality of 15 locations.

The electric tube furnace (the heater) 9 was controlled so that a temperature of the catalytic reactor 4 was maintained at 750°C. The gases to be treated comprised SiF₄, CHF₃, C₄F₈, C₄F₆, CO, C₅F₈, NF₃, SF₆ and CF₄. In this test, SiF₄ was supplied 20 at a flow rate of 60 ml/min, CHF₃ 180 ml/min, C₄F₈ 60 ml/min, C₄F₆ 10 ml/min, CO 1200 ml/min, C₅F₈ 10 ml/min, NF₃ 120 ml/min, SF₆ 120 ml/min, and CF₄ 450 ml/min. These gases were mixed with the N₂ gas supplied at a flow rate of 120 l/min and introduced 25 into the gas treatment tank 3. Air was introduced into the heating oxidative decomposing section of the gas treatment tank 3 at a flow rate of 30 l/min for supplying O₂ required for oxidation.

Similarly, pure water was introduced into the gas treatment tank 3 at a flow rate of 5 ml/min for oxidation and 30 decomposition. Table 1 shows the test results of exhaust gas treatment. In table 1, example 1 shows the results of a case

where pure water was introduced from the water heating pipe 8 whose open end is positioned at the end portion of the heating section 30, and comparative example 1 shows the results of a case where pure water was introduced from the top portion of 5 the gas treatment tank 3 (i.e., the inlet of the heating section).

Table 1

Component	Gas at the outlet of the post-treatment section	
	Example 1	Comparative example 1
CF ₄ (ppm)	<0.2	<0.2
CHF ₃ (ppm)	<0.2	<0.2
C ₄ F ₈ (ppm)	<0.2	<0.2
C ₅ F ₈ (ppm)	<0.2	<0.2
C ₄ F ₆ (ppm)	<1	<1
CO (ppm)	<2	<2
NF ₃ (ppm)	<1	<1
SF ₆ (ppm)	<1	<1
HF (ppm)	<1	<1
SiF ₄ (ppm)	<1	<1

10 It can be seen from the above results that a small amount of the gas components were detected at the outlet of the post-treatment section as shown in the example 1 and the comparative example 1. These results showed that these gas components were efficiently treated in the case of supplying 15 the pure water from the water heating pipe (water vaporizing pipe) 8 which opens at the end portion of the heating section 30 and also in the case of supplying the pure water from the

top portion of the gas treatment tank 3 (i.e., the inlet of the heating section).

Next, study results will be described of a HF gas produced by supplying water (H_2O) from the top portion of the gas treatment tank 3 (i.e., the inlet of the heating section). In order to confirm that a corrosive HF gas is produced at the heating section with a high concentration when pure water is supplied from the top portion of the gas treatment tank 3, a gas-introduction test was carried out with a testing apparatus equivalent to the above exhaust gas treatment apparatus under the following conditions.

The electric tube furnace (the heater) 9 was controlled so that a temperature of the catalytic reactor 4 was maintained at $750^{\circ}C$. In this test, CF_4 was supplied at a flow rate of 450 ml/min, CHF_3 180 ml/min, and C_4F_8 60 ml/min. These PFC gases were mixed with a N_2 gas supplied at a flow rate of 120 l/min and air for oxidation supplied at a flow rate of 30 l/min, and then introduced into the gas treatment tank 3.

In order to measure an amount of HF produced by adding water (H_2O), a gas at the outlet of the heating section 30 was sampled through the water heating pipe 8. Table 2 shows the test results of the above gas sampling. In table 2, comparative example 2 shows a case where pure water was introduced from the top portion of the gas treatment tank 3 (i.e., the inlet of the heating section), and example 2 shows a case where pure water was not introduced.

Table 2

Component	Gas at the outlet of the post-treatment section
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	Comparative example 2	Example 2
CF ₄ (ppm)	2700	3000
CHF ₃ (ppm)	<0.2	1200
C ₄ F ₈ (ppm)	<0.2	400
HF (ppm)	8000	<1

From the above test results, it can be seen that HF, which is a corrosive gas, was produced in the heating section with a high concentration in the case where water (H₂O) was added 5 from the top portion of the gas treatment tank 3 (i.e., the inlet of the heating section), as shown in comparative example 2. However, it is clear that HF was not produced in the heating section in the case where water (H₂O) was not added, as shown in example 2.

10 Next, test results will be described of a case where a H₂ gas was introduced so as to produce water (H₂O), instead of adding water (H₂O). The testing apparatus and the testing conditions such as a treatment temperature of a gas were the same as those of the above tests.

15 In this test, CF₄ was supplied at a flow rate of 450 ml/min, CHF₃ 180 ml/min, C₄F₈ 60 ml/min, SiF₄ 60 ml/min, and CO 1200 ml/min. These gases to be treated were mixed with a N₂ gas supplied at a flow rate of 120 l/min and air for oxidation supplied at a flow rate of 30 l/min, and then introduced into 20 the gas treatment tank 3.

In order to confirm a treatment effect of supplying a H₂ gas, H₂ was supplied from the water heating pipe 8 at a flow rate of 1.4 l/min, instead of supplying pure water. Table 3 shows the results of this test as example 3.

Table 3

Component	Gas at the outlet of the post-treatment section
	Example 3
CF ₄ (ppm)	<0.2
CHF ₃ (ppm)	<0.2
C ₄ F ₈ (ppm)	<0.2
CO (ppm)	<2
SiF ₄ (ppm)	<1
H ₂ (ppm)	<100

From the above test results, it can be seen that fluorine compounds were efficiently treated by introducing the H₂ gas 5 and it is possible to obtain the same treatment effect as the case of adding water (H₂O).

The exhaust gas treatment method and apparatus according to the present invention are not limited to the illustrated examples. Although certain preferred embodiments of the 10 present invention have been shown and described in detail, it should be understood that various changes and modifications may be made therein without departing from the scope of the present invention.

As described above, according to the present invention, 15 it is possible to efficiently and economically detoxify an exhaust gas containing a fluorine compound which is discharged from a semiconductor fabrication apparatus or the like without causing any damage to the heating section.

The present invention is applicable to a method and 20 apparatus for efficiently detoxifying an exhaust gas containing a fluorine compound which is discharged from a semiconductor fabrication process such as a dry-cleaning

process of an inner surface of a semiconductor fabrication apparatus or an etching process of various types of films such as oxide films.